



## Product Data Sheet

### **AMBERLITE™ FPA51 Ion Exchange Resin**

Food-grade, Macroporous, Weak Base Anion Exchange Resin

#### **Description**

AMBERLITE™ FPA51 Ion Exchange Resin is a macroporous, weakly basic, anion exchange resin containing tertiary amine functionality on a crosslinked polystyrene matrix. It has been specifically designed for the purifying liquid food streams including starch-based sweeteners and can also be used in bioprocessing applications.

#### **Nutrition Applications**

AMBERLITE™ FPA51 Ion Exchange Resin has been sized to be used both in the fixed bed system commonly used in the corn sweetener industry as well as moving bed systems and polishing mixed bed. It is suitable for the deashing, deacidification, and decolorization of glucose, fructose and related starch-based sweeteners and derivatives as well as gelatin and other food process streams such as fruit juices.

Its high level of porosity gives AMBERLITE™ FPA51 an excellent combination of physical stability and high operating efficiency resulting in long process cycle times compared to products having a higher static volume capacity. This porous network also provides a more complete adsorption and desorption of large organic molecules resulting in excellent color removal compared to other weak base anion exchange resins.

#### **Bioprocessing Applications**

A number of different antibiotic classes have been isolated, chemically modified, and used extensively by physicians in treating infectious diseases. As most traditional antibiotics were derived from yeast or bacteria, their large-scale production is based on fermentation processes. AMBERLITE™ FPA51 Ion Exchange Resin can be used for the removal of organic color bodies in those downstream bioprocesses.

#### **Applications**

- Nutrition applications
  - Sweetener deashing
  - Sweetener deacidification
  - Sweetener decolorization
- Bioprocessing applications
  - Decolorization

## Typical Properties

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### Physical Properties

Copolymer	Styrene-divinylbenzene
Matrix	Macroporous
Type	Weak base anion
Functional Group	Secondary amine ( $\geq 85\%$ )
Physical Form	Beige, opaque, spherical beads

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### Chemical Properties

Ionic Form as Shipped	Free base (FB)
Total Exchange Capacity	$\geq 1.3$ eq/L
Water Retention Capacity	56 – 62%

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### Particle Size <sup>§</sup>

Particle Diameter	490 – 690 $\mu\text{m}$
< 300 $\mu\text{m}$	$\leq 1.0\%$
> 1180 $\mu\text{m}$	$\leq 2.0\%$

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### Stability

Swelling	FB $\rightarrow$ HCl : 25%
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### Density

Shipping Weight	660 g/L
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<sup>§</sup> For additional particle size information, please refer to the [Particle Size Distribution Cross Reference Chart](#) (Form No. 177-01775).

## Suggested Operating Conditions

Maximum Operating Temperature	100°C (212°F)
Bed Depth, min.	700 mm (2.3 ft)
Flowrates	
Service	2 – 6 BV*/h (depending on syrup concentration)
Backwash	See Figure 1
Regeneration	1 – 2 BV/h
Slow Rinse	Regeneration flowrate for 2 BV
Fast Rinse (if applicable)	Service flowrate for 5 – 10 BV (with condensate or softened water)
Contact Time	
Regeneration	≥ 30 – 45 minutes
Regenerant	
Concentration	4%
Level	60 kg/m <sup>3</sup> (3.8 lb/ft <sup>3</sup> )

\* 1 BV (Bed Volume) = 1 m<sup>3</sup> solution per m<sup>3</sup> resin or 7.5 gal per ft<sup>3</sup> resin

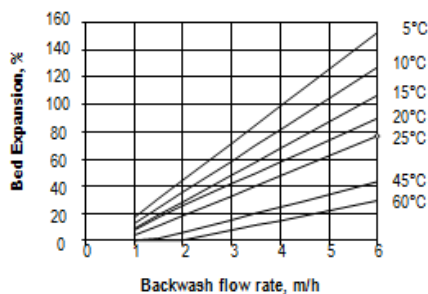
## Hydraulic Characteristics

Estimated bed expansion of AMBERLITE™ FPA51 Ion Exchange Resin as a function of backwash flowrate and temperature is shown in Figure 1.

Estimated pressure drop for AMBERLITE™ FPA51 as a function of service flowrate and viscosity is shown in Figure 2. These pressure drop expectations are valid at the start of the service run with clean feed and a well-classified bed.

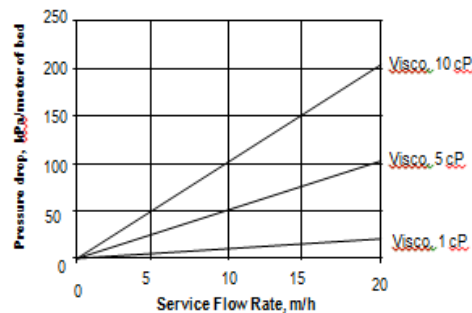
**Figure 1: Backwash Expansion**

Temperature = 5 – 60°C (41 – 140°F)



**Figure 2: Pressure Drop**

Viscosity = 1 – 10 cP



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Please be aware of the following:

- **WARNING:** Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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