



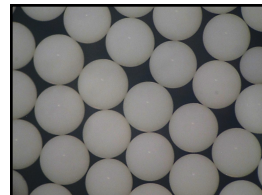
Product Data Sheet

AmberLite™ HPR900 SO₄ Ion Exchange Resin

Macroporous, Strong Base Anion Exchange Resin for Condensate Polishing for the Power Industry and Industrial Demineralization Applications

Description

AmberLite™ HPR900 SO₄ Ion Exchange Resin is specifically designed for use in condensate polishing beds at fossil-fired electric generating stations and industrial demineralization applications when a balance of operating performance, simple operation, long resin life, and cost-effective operation is required.



The macroporous structure of AmberLite™ HPR900 SO₄ provides resistance to surface fouling as well as physical, osmotic, and oxidative stresses, which allows increased resin lifetime in operation. The resin can operate reliably under the high flowrate and pressure drop conditions that are typically used in condensate polishers.

This resin is designed to be used in combination with AmberLite™ HPR252 H Ion Exchange Resin and AmberLite™ 600i Inert Resin in TRIOBED™ Condensate Polishers, providing an optimized balance of stability, operating capacity, low pressure drop, and regeneration efficiency.

When high water quality and long runtime are needed, AmberLite™ HPR1300 H Ion Exchange Resin is a trusted choice.

When compliance with the China National Standard specifications for fossil power condensate polishing applications, including the China Strong Osmotic Ball Mill test, is important, AmberLite™ HPR2800 H Ion Exchange Resin is the recommended cation pair since both resins are compliant with the standard.

Resin Pairings

Recommended pairing:

- AmberLite™ HPR252 H Ion Exchange Resin (macroporous)
- AmberLite™ HPR1300 H Ion Exchange Resin (gel)
- AmberLite™ HPR2800 H Ion Exchange Resin (macroporous)

Applications

- Mixed bed condensate polishing in fossil power plants
- Mixed bed polishing in industrial demineralization
- Systems requiring exceptionally high osmotic stability

Historical Reference

AmberLite™ HPR900 SO₄ Ion Exchange Resin has previously been sold as AmberSep™ 900 SO₄ Ion Exchange Resin.

Typical Properties

Physical Properties	
Copolymer	Styrene-divinylbenzene
Matrix	Macroporous
Type	Strong base anion
Functional Group	Trimethylammonium
Physical Form	White, opaque, spherical beads
Chemical Properties	
Ionic Form as Shipped	SO ₄ ²⁻
Total Exchange Capacity	≥ 1.0 eq/L (Cl ⁻ form)
Water Retention Capacity	60.0 – 68.0% (Cl ⁻ form)
Particle Size §	
Particle Diameter	500 – 700 µm
Uniformity Coefficient	≤ 1.45
< 300 µm	≤ 0.5%
> 1180 µm	≤ 1.0%
Stability	
Whole Uncracked Beads	≥ 95%
Strong Osmotic Ball Mill Test	≥ 92%
Swelling	Cl ⁻ → OH ⁻ ≤ 25% SO ₄ ²⁻ → OH ⁻ ≤ 15%
Density	
Particle Density	1.09 g/mL
Shipping Weight	695 g/L

§ For additional particle size information, please refer to the [Particle Size Distribution Cross Reference Chart](#) (Form No. 45-D00954-en).

Suggested Operating Conditions

Temperature Range (OH ⁻ form) ‡	5 – 100°C (41 – 212°F)
pH Range (Stable)	0 – 14

‡ Operating at elevated temperatures, for example above 60 – 70°C (140 – 158°F), may impact the purity of the loop and resin life. Contact our technical representative for details.

For additional information regarding recommended minimum bed depth, operating conditions, and regeneration conditions for [mixed beds](#) (Form No. 45-D01127-en) or [separate beds](#) (Form No. 45-D01131-en) in water treatment, please refer to our Tech Facts.

Hydraulic Characteristics

Estimated bed expansion of AmberLite™ HPR900 SO₄ Ion Exchange Resin as a function of backwash flowrate and temperature is shown in Figure 1.

Estimated pressure drop for AmberLite™ HPR900 SO₄ as a function of service flowrate and temperature is shown in Figure 2. These pressure drop expectations are valid at the start of the service run with clean water and a well-classified bed.

Figure 1: Backwash Expansion

Temperature = 10 – 60°C (50 – 140°F)

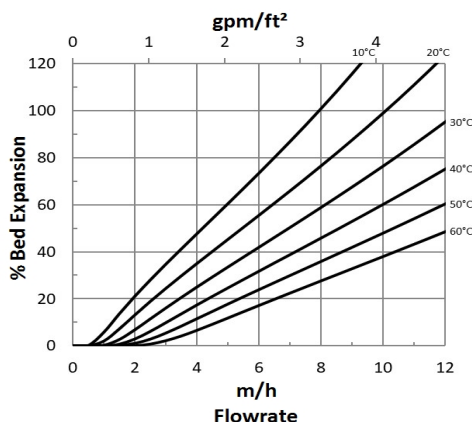
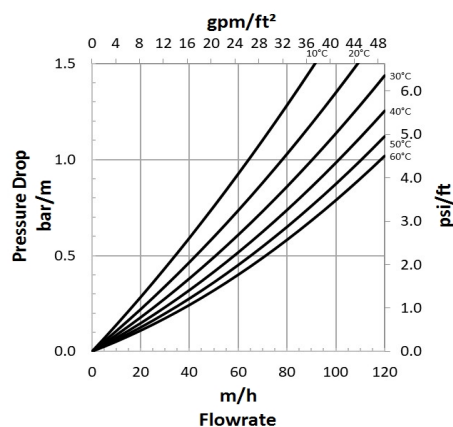


Figure 2: Pressure Drop

Temperature = 10 – 60°C (50 – 140°F)



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Please be aware of the following:

- **WARNING:** Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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