

Product Data Sheet

AMBERSEP™ 400 SO₄ Ion Exchange Resin

Industrial-grade Strong Base Anion Exchanger

Description

AMBERSEP™ 400 SO₄ Ion Exchange Resin is a gel, Type I strong basic anion exchange resin with outstanding performance for uranium recovery. Its excellent selectivity for the uranyl sulfate ion over other anions, high operating capacity, excellent mechanical and physical stability, and its resistance to fouling make it the resin of choice. AMBERSEP™ 400 SO₄ is well-suited for the recovery of uranium from sulfuric acid leach systems using fixed beds, in situ leaching, fluidized beds, or Resin In Pulp (RIP) applications.

AMBERSEP™ 400 SO₄ is supplied in the sulfate form in order to minimize the presence of chloride upon start-up.

If used in sulfuric acid leach systems, no preconditioning of this resin is required and the resin can be used as supplied.

Applications

- Uranium extraction from sulfuric acid leach systems
 - Fixed beds
 - In situ leaching
 - Fluidized beds
 - Resin In Pulp (RIP) systems

Typical Properties

Styrene-divinylbenzene
Gel
Strong base anion, Type I
Trimethylammonium
Amber, translucent, spherical beads
SO ₄ ²⁻
≥ 1.40 eq/L (Cl⁻ form)
40 − 47% (Cl ⁻ form)
600 – 750 μm
≤ 1.60
≤ 1.0%
≤ 5.0%
730 g/L

[§] For additional particle size information, please refer to the Particle Size Distribution Cross Reference Chart (Form No. 177-01775).

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Product Stewardship

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DuPont strongly encourages its customers to review both their manufacturing processes and their applications of DuPont products from the standpoint of human health and environmental quality to ensure that DuPont products are not used in ways for which they are not intended or tested. DuPont personnel are available to answer your questions and to provide reasonable technical support. DuPont product literature, including safety data sheets, should be consulted prior to use of DuPont products. Current safety data sheets are available from DuPont.

Please be aware of the following:

 WARNING: Oxidizing agents such as nitric acid attack organic ion exchange resins under certain conditions. This could lead to anything from slight resin degradation to a violent exothermic reaction (explosion). Before using strong oxidizing agents, consult sources knowledgeable in handling such materials.

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